



## PROSIMPLUS APPLICATION EXAMPLE

### NAPHTHALENE SEPARATION

#### INTEREST OF THIS EXAMPLE

This example illustrates a process to purify naphthalene from a mixture containing 14 components in a three columns distillation train. This example mainly focuses on two-phase (vapor-liquid) distillation columns. For each of the three distillation columns, several specifications are set on the output streams, illustrating the way to set "non-standard" specifications in the multi-stage separation modules of ProSimPlus.

This example also demonstrates the way to use modules for measurement of some streams parameters and also how to handle the "information streams" in order to determine the global naphthalene recovery ratio between process input and output.

ACCESS	<input checked="" type="checkbox"/> Free-Internet	<input type="checkbox"/> Restricted to ProSim clients	<input type="checkbox"/> Restricted	<input type="checkbox"/> Confidential
--------	---	---	-------------------------------------	---------------------------------------

CORRESPONDING PROSIMPLUS FILE	<a href="#">PSPS_E04_EN - Naphthalene Separation.pmp3</a>
-------------------------------	---

*Reader is reminded that this use case is only an example and should not be used for other purposes. Although this example is based on actual case it may not be considered as typical nor are the data used always the most accurate available. ProSim shall have no responsibility or liability for damages arising out of or related to the use of the results of calculations based on this example.*

## TABLE OF CONTENTS

<b>1. PROCESS MODELING</b>	<b>3</b>
1.1. Process description	3
1.2. Process flowsheet	3
1.3. Specification	4
1.4. Components	4
1.5. Thermodynamic model	4
1.6. Operating conditions	5
1.7. "Hints and Tips"	7
<b>2. RESULTS</b>	<b>8</b>
2.1. Comments on results	8
2.2. Mass and energy balances	8
2.3. Column C101 composition profiles	10
2.4. Column C102 composition profiles	10
2.5. Column C103 composition profiles	11
<b>3. REFERENCES</b>	<b>11</b>

# 1. PROCESS MODELING

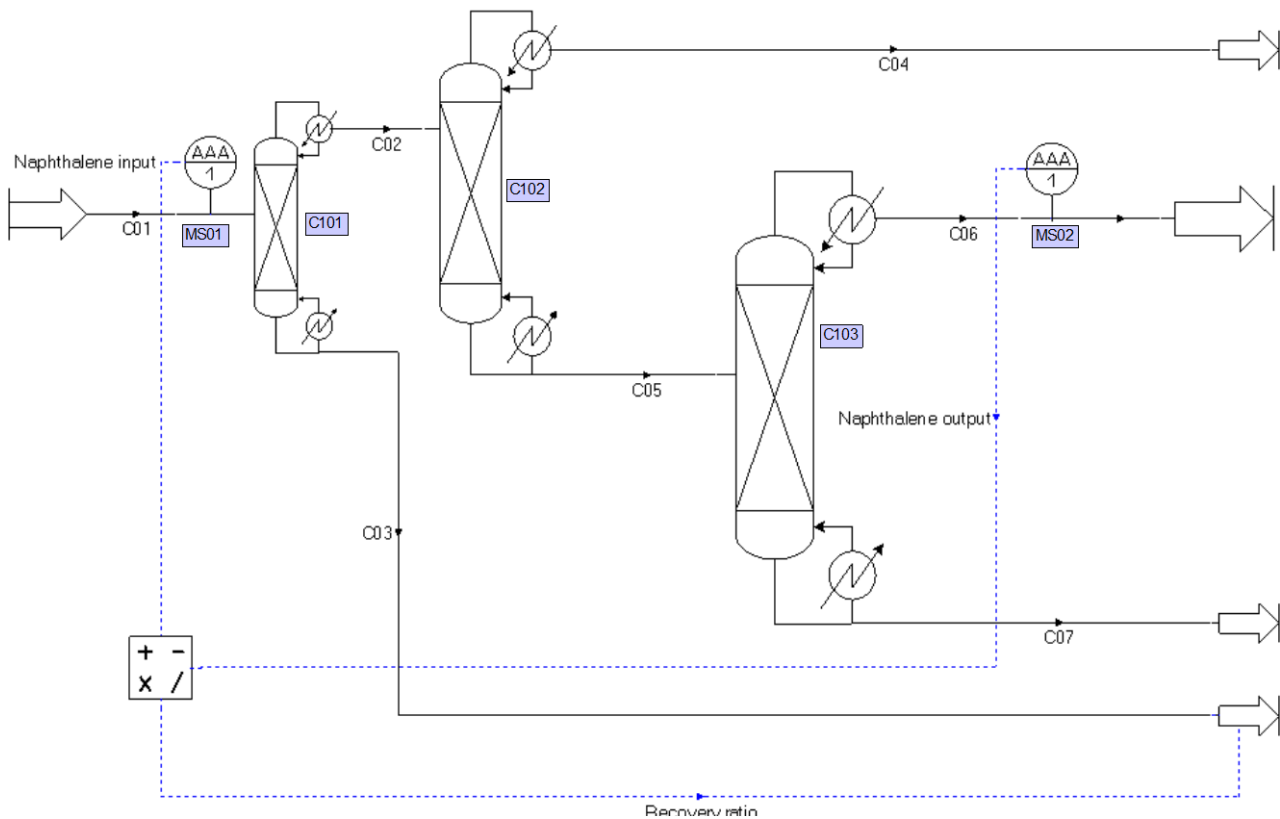
## 1.1. Process description

The objective of this process is to produce a naphthalene stream with a 97 weight % purity by distillation in a three distillation columns train.

The first column removes Anthracene into the bottom product. The overhead from the first column contains the bulk of the feed and is fed to the second column where the 1,2,4-Trimethylbenzene is removed. A small loss of Naphthalene is allowed in this column. Finally, the last column produces the high purity Naphthalene.

This example is taken from [1].

## 1.2. Process flowsheet



*Naphthalene separation process flowsheet*

### 1.3. Specification

The process specification is to recover a Naphthalene stream with a purity of 97 % wt.

### 1.4. Components

Components taken into account in the simulation are of two kinds:

- A first set of components are taken from the ProSim standard database, provided with ProSim's software. These components are defined in the table below:

- ❖ Naphthalene
- ❖ 2,3-Xylenol
- ❖ 2-Methylnaphtalene
- ❖ 1,2,4-Trimethylbenzene
- ❖ 2,4-Xylenol
- ❖ Fluorene
- ❖ Indene
- ❖ n-Dodecane
- ❖ Anthracene
- ❖ m-Cresol

- The second set of components for which all the required properties are automatically generated with the pseudo-compound generation service, from the knowledge of the following properties:

Name	Molecular weight (g/mol)	Density	NBP (°C)
Diphenyl	150.425	1.0076	254.6
23 Di Methyl Naph	152.817	1.0407	268.0
Acet. Naph.	152.027	1.0930	279.8
Di BenzoF	164.451	0.9947	274.6

### 1.5. Thermodynamic model

Given the nature of the compounds involved in the process, the thermodynamic model used to represent phase equilibria and the enthalpy calculations is the Peng-Robinson (PR) [2] cubic equation of state, without binary interaction parameters.

## 1.6. Operating conditions

✓ Process feed

Temperature (°C)	100
Pressure (mmHg)	780
<i>Mass partial flowrates (kg/h)</i>	
1,2,4-Trimethylbenzene	376.00
Indene	101.00
m-Cresol	5.50
2,4-Xylenol	9.60
n-dodecane	13.20
2,3-Xylenol	5.10
Naphthalene	2266.00
Diphenyl	62.30
23 Di Methyl Naph	78.90
2-Methyl Naphthalene	55.90
Acet. Naph.	72.10
Di Benzof	90.60
Fluorene	110.50
Anthracene	213.00

## ✓ Column C101

<i>Operating parameters</i>	<i>Value</i>
Type of column	Two-phase distillation column
Number of theoretical stages	7
Feed stage	5
Pressure (mmHg)    Top	760
Stage 2	780
Reflux molar flowrate (kmol/h)	9 (initialization)
Liquid distillate molar flowrate (kmol/h)	23.9 (initialization)

Additional specifications for column C101:

<i>Specifications</i>	<i>Product type</i>	<i>Component</i>	<i>Value</i>	<i>Phase</i>	<i>Type</i>	<i>Action</i>
1: Recovery ratio	Vapor distillate	Naphthalene	0.993	Vap.	Mol.	Vapor distillate flowrate
2: Recovery ratio	Bottom liquid product	Anthracene	0.995	Liq.	Mol.	Reflux flowrate

## ✓ Column C102

<i>Operating parameters</i>	<i>Value</i>
Type of column	Two-phase distillation column
Number of theoretical stages	14
Feed stage	3
Pressure (mmHg)    Top	760
Stage 2	780
Reflux molar flowrate (kmol/h)	43 (initialization)
Liquid distillate molar flowrate (kmol/h)	4.3054 (initialization)

Additional specifications for column C102:

<i>Specifications</i>	<i>Product type</i>	<i>Components</i>	<i>Value</i>	<i>Phase</i>	<i>Type</i>	<i>Action</i>
1: Purity	Vapor distillate	Naphthalene	0.15	Vap.	Mass.	Vapor distillate flowrate
2: Recovery ratio	Bottom liquid product	1,2,4-Trimethylbenzene	0.01	Liq.	Mol.	Reflux flowrate



## 2. RESULTS

### 2.1. Comments on results

The convergence of the three columns is reached in very few calculations. The output stream C06 effectively contains Naphthalene with a purity of 97% wt. The global recovery ratio of Naphthalene is calculated to be 95.6%.

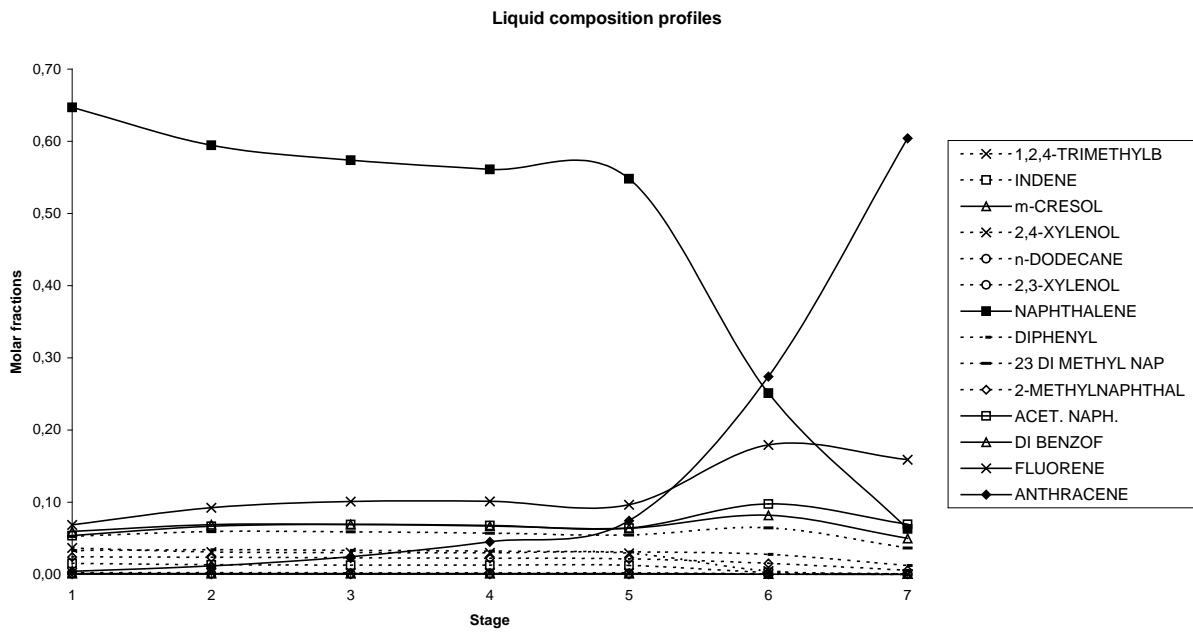
### 2.2. Mass and energy balances

Streams		C01	C02	C03	C04
From		>>>	C101	C101	C102
To		C101	C102	>>>	>>>
Partial flowrates		kg/h	kg/h	kg/h	kg/h
1,2,4-TRIMETHYLBENZENE		376.00	375.89	0.11	372.13
INDENE		101.00	100.91	0.09	60.66
m-CRESOL		5.50	5.49	0.01	0.86
2,4-XYLENOL		9.60	9.57	0.03	0.64
n-DODECANE		13.20	13.20	0.00	11.00
2,3-XYLENOL		5.10	5.08	0.02	0.25
NAPHTHALENE		2266.00	2250.14	15.86	78.70
DIPHENYL		62.30	58.74	3.56	0.08
23 DI METHYL NAPH		78.90	67.94	10.96	0.03
2-METHYLNAPHTHALENE		55.90	54.34	1.56	0.25
ACET. NAPH.		72.10	51.40	20.70	9.85E-03
DI BENZOF		90.60	74.50	16.10	2.04E-02
FLUORENE		110.50	58.58	51.92	8.37E-03
ANTHRACENE		213.00	1.07	211.93	1.31E-06
Total flowrate	kg/h	3459.70	3126.84	332.86	524.64
Mass fractions					
1,2,4-TRIMETHYLBENZENE		0.1087	0.1202	0.0003	0.7093
INDENE		0.0292	0.0323	0.0003	0.1156
m-CRESOL		0.0016	0.0018	3.18E-05	0.0016
2,4-XYLENOL		0.0028	0.0031	0.0001	0.0012
n-DODECANE		0.0038	0.0042	1.31E-05	0.0210
2,3-XYLENOL		0.0015	0.0016	0.0001	0.0005
NAPHTHALENE		0.6550	0.7196	0.0477	0.15
DIPHENYL		0.0180	0.0188	0.0107	0.0002
23 DI METHYL NAPH		0.0228	0.0217	0.0329	6.07E-05
2-METHYLNAPHTHALENE		0.0162	0.0174	0.0047	0.0005
ACET. NAPH.		0.0208	0.0164	0.0622	1.88E-05
DI BENZOF		0.0262	0.0238	0.0484	3.89E-05
FLUORENE		0.0319	0.0187	0.1560	1.60E-05
ANTHRACENE		0.0616	0.0003	0.6367	2.50E-09
Physical state		Liquid	Vapor	Liquid	Vapor
Temperature	°C	100.00	223.34	299.18	181.89
Pressure	mmHg	780.00	760.00	780.00	760.00
Enthalpy	kcal/h	-242963.12	209512.39	6224.06	29454.78
Vapour fraction		0.00	1.00	0.00	1.00

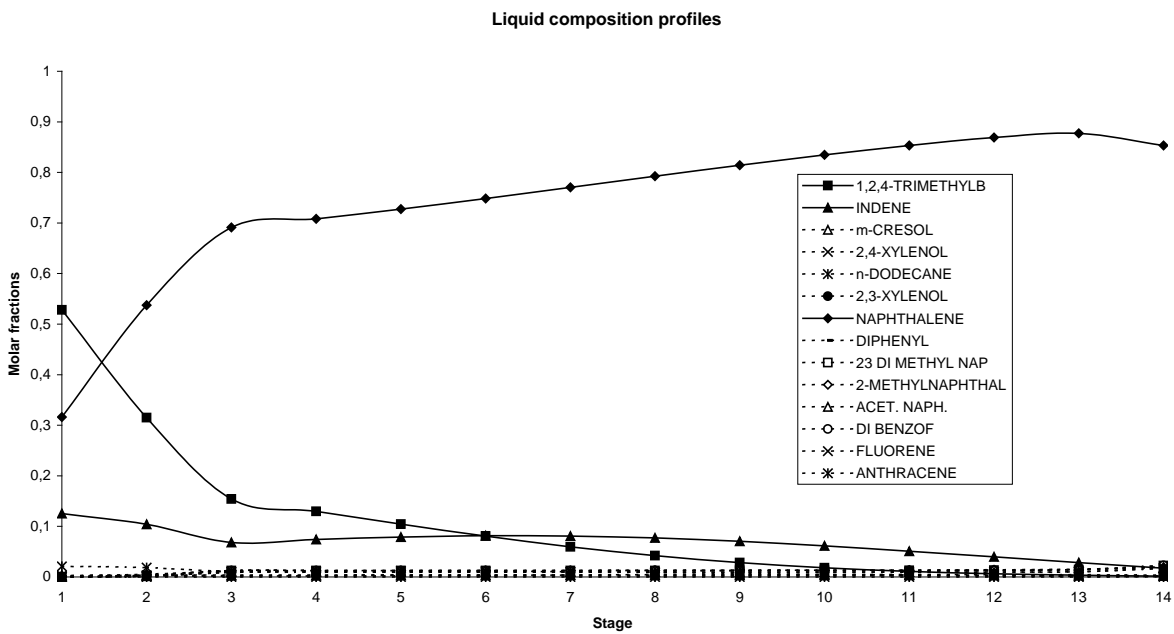


Streams		C05	C06	C07
From		C102	C103	C103
To		C103	>>>	>>
Partial flowrates		kg/h	kg/h	kg/h
1,2,4-TRIMETHYLBENZENE		3.76	3.76	3.89E-08
INDENE		40.26	40.26	2.12E-05
m-CRESOL		4.63	4.63	5.70E-05
2,4-XYLENOL		8.93	8.93	9.32E-04
n-DODECANE		2.20	2.20	1.33E-07
2,3-XYLENOL		4.82	4.82	1.61E-03
NAPHTHALENE		2171.44	2166.01	5.43
DIPHENYL		58.66	1.94E-03	58.65
23 DI METHYL NAPH		67.91	7.30E-06	67.91
2-METHYLNAPHTHALENE		54.09	7.82E-01	53.31
ACET. NAPH.		51.39	0	51.39
DI BENZOF		74.48	1.69E-06	74.48
FLUORENE		58.57	0	58.57
ANTHRACENE		1.06	0	1.07
Total flowrate	kg/h	2602.20	2231.39	370.81
Mass fractions				
1,2,4-TRIMETHYLBENZENE		0.0014	0.0017	1.05E-10
INDENE		0.0155	0.0180	5.72E-08
m-CRESOL		0.0018	0.0021	1.54E-07
2,4-XYLENOL		0.0034	0.0040	2.51E-06
n-DODECANE		0.0008	0.0010	3.58E-10
2,3-XYLENOL		0.0019	0.0022	4.35E-06
NAPHTHALENE		0.8345	0.9707	0.0146
DIPHENYL		0.0225	8.70E-07	0.1582
23 DI METHYL NAPH		0.0261	3.27E-09	0.1831
2-METHYLNAPHTHALENE		0.0208	3.51E-04	0.1438
ACET. NAPH.		0.0197	0	0.1386
DI BENZOF		0.0286	7.58E-10	0.2009
FLUORENE		0.0225	0	0.1580
ANTHRACENE		0.0004	0	0.0029
Physical state		Liquid	Vapor	Liquid
Temperature	°C	221.62	218.46	265.16
Pressure	mmHg	780.00	780.00	780.00
Enthalpy	kcal/h	-42553.90	141503.76	1940.57
Vapour fraction		0.00	1.00	0.00

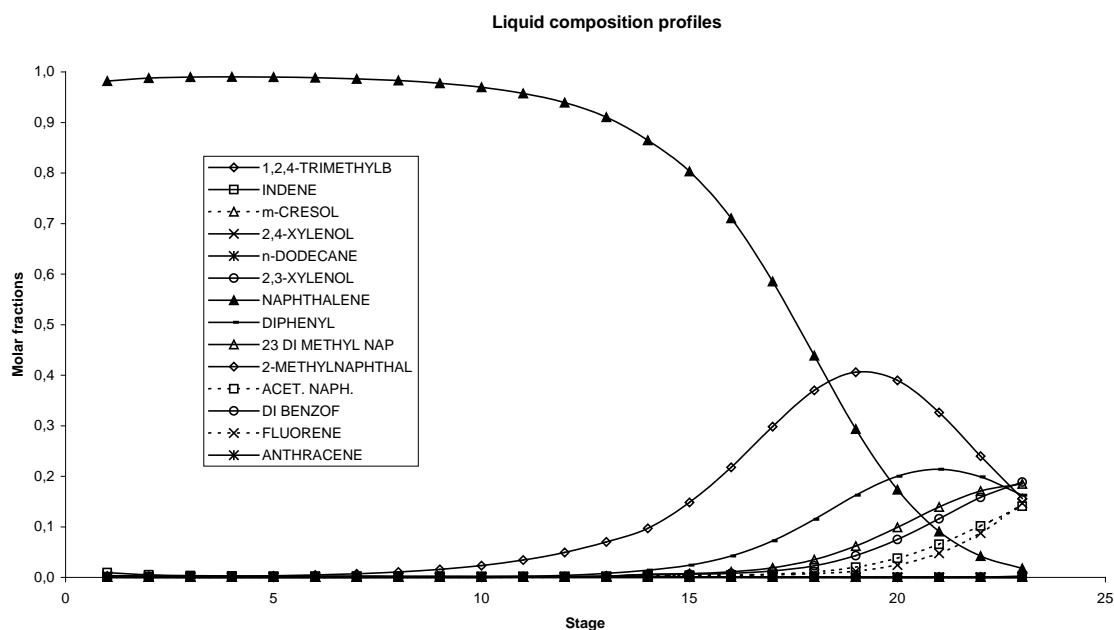
### 2.3. Column C101 composition profiles



### 2.4. Column C102 composition profiles



## 2.5. Column C103 composition profiles



## 3. REFERENCES

- [1] Application Briefs  
Process Simulation Program  
Simulation Sciences Inc.  
Revision 1, Jan 1983
- [2] Peng Y.D., Robinson D.B.  
"A new two constant equation of state"  
I.E.C. Fundam., 15, 1, 59-64 (1976)