

# Sea Water Modeling with Simulis® Thermodynamics

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ProSim



## 1 Simulis® Thermodynamics: a Thermophysical calculation server

Computes thermophysical properties and phase equilibria on pure components or mixtures:

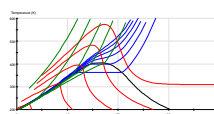
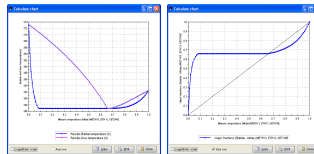
### PROPERTIES CALCULATED

- |                                   |   |
|-----------------------------------|---|
| <b>Transport properties</b>       | <b>Liquid-Vapor equilibria</b>                            |
| Isobaric specific heat (Cp)       | Bubble and dew temperatures and pressures                 |
| Dynamic viscosity                 | Flash at given temperature (T) and pressure (P)           |
| Thermal conductivity              | Flash at given vaporization ratio and P (or T)            |
| Density                           | Flash at given enthalpy (H) and P (or T, or V, or U)      |
| Molar Volume                      | Flash at given entropy (S) and P (or T, or V, or H, or U) |
| Molar density                     | Flash at given internal energy (U) and P (or T, or V)     |
| Surface tension                   | Flash at given volume (V) and P (or T)                    |
| Molecular weight                  | Phase Envelope  |
| <b>Compressibility properties</b> | <b>Liquid-Liquid equilibria</b>                           |
| Compressibility factor            | Flash at given temperature and pressure                   |
| Gamma (Cp/Cv ratio)               | <b>Liquid-Liquid-Vapor equilibria</b>                     |
| Sound speed                       | Bubble temperature  |
| <b>Thermodynamic properties</b>   | Flash at given enthalpy and pressure                      |
| Enthalpy (H)                      | Flash at given temperature and pressure                   |
| Entropy (S)                       | Flash at given vaporization ratio and pressure            |
| Internal energy (U)               | <b>Non-ideal properties</b>                               |
| Isochoric specific heat (Cv)      | Activity coefficients                                     |
| Enthalpy of vaporization          | Fugacity coefficients and Fugacity                        |

Derivatives of the properties with respect to temperature, pressure and number of moles are also provided

### THERMODYNAMIC MODELS AVAILABLE

- Equations of State**
- Soave-Redlich-Kwong (SRK)
  - Peng-Robinson (PR)
  - Lee-Kesler-Plöcker (LKP)
  - Predictive Peng-Robinson 78 (PPR78)
  - Nakamura
  - etc....
- Activity coefficients models**
- NRTL
  - UNIQUAC
  - UNIFAC (Larsen, Dortmund,...)
  - Wilson
  - etc....
- Specific systems**
- Pure Water
  - Amines
  - Sour-Water
  - etc....
- Combined approach models**
- MHV2
  - MHV1
  - PSRK
  - etc....
- Electrolytes**
- Edwards
  - UNIQUAC electrolyte
  - ULPDHS
  - etc....



The various available methods can be combined in order to configure a thermodynamic model adapted to a specific system

Simulis® is the name of the new component oriented software suite of ProSim

### A full set of services available:

- Data regression of experimental properties
- Graphical display of properties on temperature, pressure or composition ranges
- Generation of property tables
- Export of PSF files (HTFS), PVT files (OLGA)
- Estimation of pure component properties
- Plot of phase envelope diagrams
- Residue curves calculation & ternary diagrams
- Calculation of petroleum fractions properties
- Unit conversions
- UNIFAC models manager
- etc....

All these services become automatically available in your usual software since it integrates Simulis® Thermodynamics

### Uses the widely validated thermodynamic library of ProSim:

- Maturity of the architecture
- Reliability of the results
- Robustness of algorithms

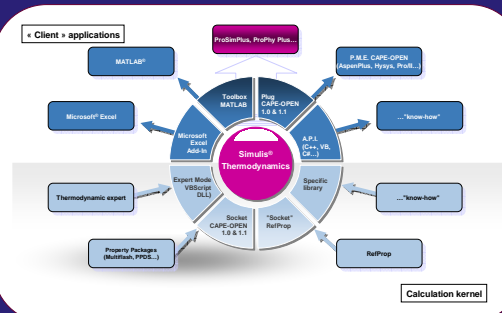
Supplied with a database of over 1 900 components including AIChE's DIPPR® database and access to your "private" databases of pure components properties.



### Other software components included:

- Simulis® Conversions: physical units conversion management tool
- Simulis® Properties: pure substances properties server

## 2 Easy to integrate and to be integrated



Any application that integrates Simulis® Thermodynamics automatically inherits from its CAPE-OPEN standard compliance

## 3 Expert mode to add your own thermodynamics models (new or existing ones)

Dynamic Link Library (DLL)



VBScript

### Capability to use legacy codes

End-users can introduce their own know-how within Simulis® Thermodynamics

Ability to merge native ProSim codes and legacy codes.

Ability to use native pure compound properties in legacy codes.

Tests and debugging facilities supplied to developers

## 4 Application example: Saline systems and Sea Water Modeling

Inflow (moles)	Outflow (moles of major species)		
H <sub>2</sub> O	55.508	Aqueous Phase	
NaCl	1.50	H <sub>2</sub> O	53.610
CaSO <sub>4</sub>	1.00	Na <sup>+</sup>	1.50
		MgCl <sub>2</sub>	1.50
		Ca <sup>2+</sup>	0.051
		SO <sub>4</sub> <sup>2-</sup>	0.051
		Solid phase	
		CaSO <sub>4</sub> ·2H <sub>2</sub> O	0.949

Table 4: System Gypsum – Sodium Chloride – Water at 25°C: mass results

Component	Molality (mol.kg <sup>-1</sup> )
NaCl	0.424
MgCl <sub>2</sub>	0.0553
Na <sub>2</sub> SO <sub>4</sub>	0.0291
CaCl <sub>2</sub>	0.0105
KCl	0.0094

Table 1: artificial seawater composition [1][2]

T(°C)	Ionic strength I	γ <sub>NaCl</sub> Experiment [1]	γ <sub>NaCl</sub> Calc.	γ <sub>Na<sub>2</sub>SO<sub>4</sub></sub> Experiment [2]	γ <sub>Na<sub>2</sub>SO<sub>4</sub></sub> Calc.
25	0.718	0.672 ± 0.007	0.666	0.378 ± 0.016	0.3735

Table 2: modelled values and measurements in artificial sea water

Reference	Component					
	γ <sub>Na<sup>+</sup></sub>	γ <sub>Mg<sup>2+</sup></sub>	γ <sub>Ca<sup>2+</sup></sub>	γ <sub>K<sup>+</sup></sub>	γ <sub>Cl<sup>-</sup></sub>	γ <sub>SO<sub>4</sub><sup>2-</sup></sub>
Experimental	0.67 [3]	0.26 [4]	0.21 [5]	0.61 [6]	0.68 [1]	0.11 [2]
This work	0.672	0.258	0.212	0.610	0.681	0.114

Table 3: Major Ion Activity coefficients in Seawater (T = 25°C and S<sup>0</sup><sub>00</sub> = 35)

[1] R.F. Platford, J. Mar. Res., 23 (1965) 5; [2] R.F. Platford and T. Daffoe, J. Mar. Res., 23 (1965) 63; [3] R.F. Platford, J. Fish. Res. Bd. Can., 22 (1965) 885; [4] M.E. Thompson, Science, 153 (1964) 1643; [5] M.E. Thompson and J.W. Ross, J. Mar. Res., 23 (1965) 63; [6] G. Mangelsdorf Jr. and T.R.S. Wilson, J. Phys. Chem., 75 (1971) 1418