Getting started with BatchReactor®

Use Case 2: Simulation of bioreactions

Software & Services In Process Simulation

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The challenge tied to the modeling of bioreactions results from the complexity of bioreaction schemes and kinetics. With the advanced mode available in Simulis Reactions, the user can import libraries of bioreaction kinetic models that can be easily modified and adjusted to a wide range of bioreactions.

As an example, this document showcases how to simulate the classical rate of biomass growth using the model of Monod.

Here are the steps to follow:

- Step 1: Selection of the compounds
- Step 2: Configuration of the bioreaction
- Step 3: Specification of the equipment and the operating scenario

Description of the model

A bioreaction corresponds to an autocatalyzed reaction, since the biomass has both roles of catalyst and product of the reaction:

 $Substrates + Biomass \rightarrow More Biomass + Products$

The bioreaction stoichiometry can be described as follows:

$$\sum_{i=1}^{NC} v_i S_i \xrightarrow{X} \sum_{i=1}^{NC} v_i' P_i$$

With:

NC Number of components

S Substrates (i.e., glucose, oxygen and other limiting or non-limiting substrates...)

P Products (i.e., biomass growth, products of interest and other by-products)

X Biomass (e.g., micro-organisms catalysing the bioreaction)

 v_i, v_i stoichiometric coefficients (positive value for products and negative value for substrates)

The global bioreaction rate can be defined as follows:

$$r_G = \frac{1}{\nu_i} r_{S_i} = \frac{1}{\nu_{i'}} r_{P_i}$$

With:

r_G Global reaction rate of the bioreaction (positive value)

r_{Si} Specific rate of consumption of the substrate S_i (negative value)

r_{pi} Specific rate of formation of the product P_i (positive value)

Using the advanced mode available in Simulis Reactions, the user can import a library of bioreaction kinetic models. Two different formalisms are available, where it is possible to combine elementary kinetic models $(r(C_{Si}))$ to take into account multiple different uptake/inhibition kinetics:

Product of elementary kinetic models (*Option* 1):

$$r_G = \left(\alpha. \mu_{max} \prod_{i=1}^{NLS} r(C_{Si}) + \beta\right). C_X$$

Sum of elementary kinetic models (**Option 2**):

$$r_G = \left(\alpha.\sum_{i=1}^{NLS} \mu_{max,i}. \boldsymbol{r}(\boldsymbol{C_{Si}}) + \beta\right). C_X$$

With:

 $\begin{array}{ccc} \alpha & & \text{The growth-related coefficient} \\ \beta & & \text{The non-growth-related coefficient} \end{array}$

 μ_{max} The maximum growth rate

 C_{Si} , C_X Substrate, inhibitor or biomass concentration

The elementary kinetic models $r(C_{Si})$ are selected from a standard list that can be enriched by the user



Model index	Description	Equation of the term $r(\mathcal{C}_{\mathcal{S}i})$
1	Monod	$\frac{C_S}{K_S + C_S}$
2	Hill	$\frac{C_S^N}{K_S^N + C_S^N}$
3	Contois	$\frac{C_S}{K_S C_X + C_S}$
Etc		

Description of the model

APPLICATION EXAMPLE: this simple application example is based on the modeling of biomass growth, corresponding to the following equation:

 $Substrate + Biomass \rightarrow More Biomass$

The bioreaction stoichiometry can be described as follows: $S \xrightarrow{X} X$

The global rate of bioreaction is represented by the model of Monod:

$$r_G = \mu. C_X = \mu_{max} \left(\frac{C_S}{K_S + C_S} \right) C_X$$

With:

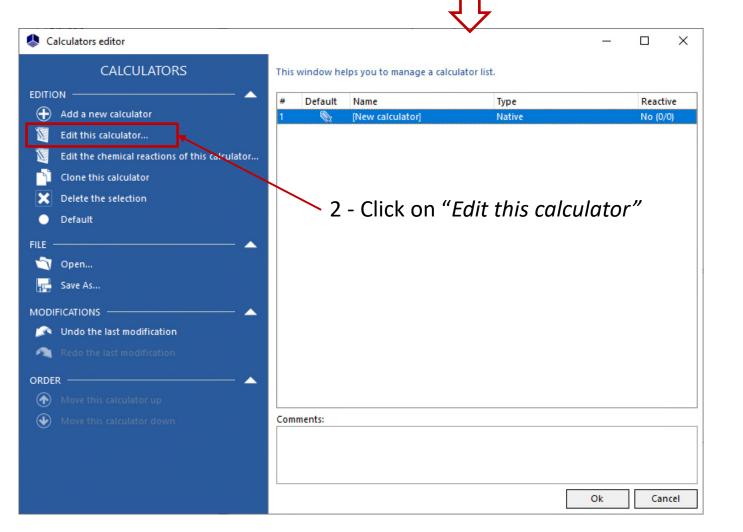
Model parameters	Definition	Value
μ_{max}	Maximum growth rate	4.10 ⁻⁵ s ⁻¹
K_S	Saturation constant	2.8 g/L
C_S , C_X	Concentration of substrate (S) and biomass (X)	Process variables



Step 1: Selection of the compounds

1 - Click on the "Edit the thermodynamics and compounds" icon to access the "Calculators editor" window





Step 1: Selection of the compounds

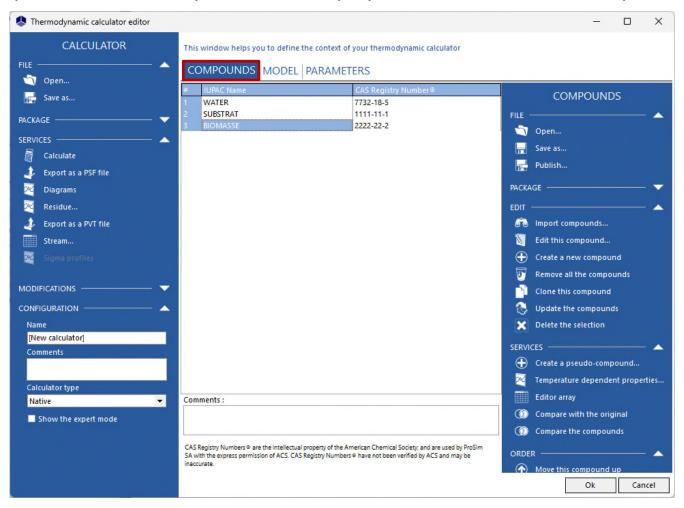
This application example requires to import the following compounds:

- WATER
- SUBSTRATE (modeled as "Glucose"):
 - Import of the "Glucose" from the compounds database
 - Modification of the name: SUBSTRATE
 - Modification of the CAS number(*): 1111-11-1
- BIOMASS (modeled as "Glucose"):
 - Import of the "Glucose" from the compounds database
 - Modification of the name: BIOMASS
 - Modification of the CAS number^(*): 2222-22-2

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Step 1: Selection of the compounds

Once this step is done, the compounds are displayed as follows in the "Compounds" tab:

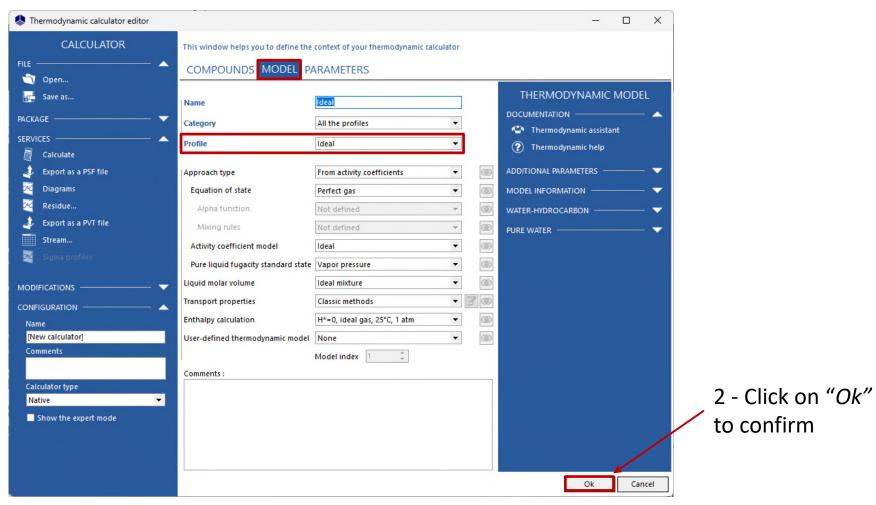




For more information about compounds selection, please refer to "Getting started with Simulis Thermodynamics, use case 1"

Step 1: Selection of the compounds

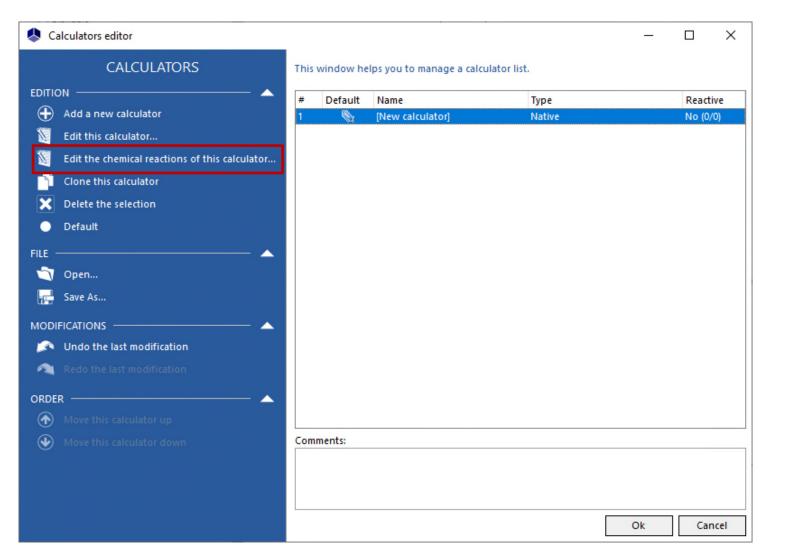
1 - In the "Model" tab, select the "ideal" thermodynamic profile:



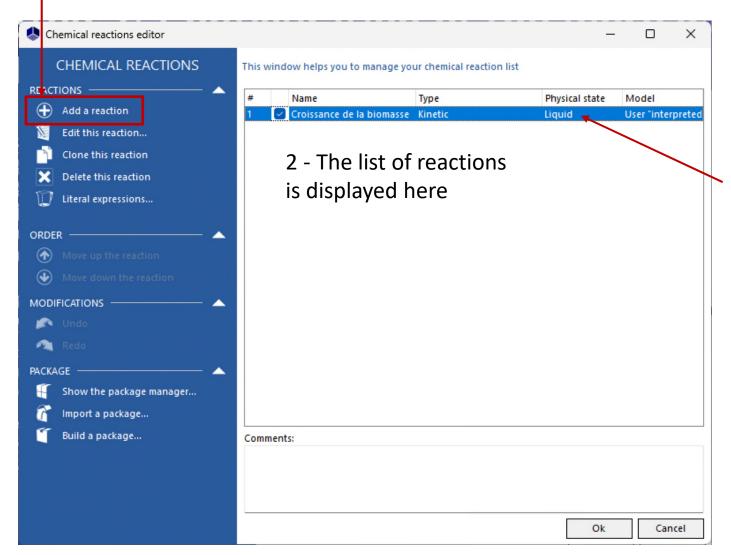


For more information about thermodynamic model configuration, please refer to "Getting started with Simulis Thermodynamics, use case 1"

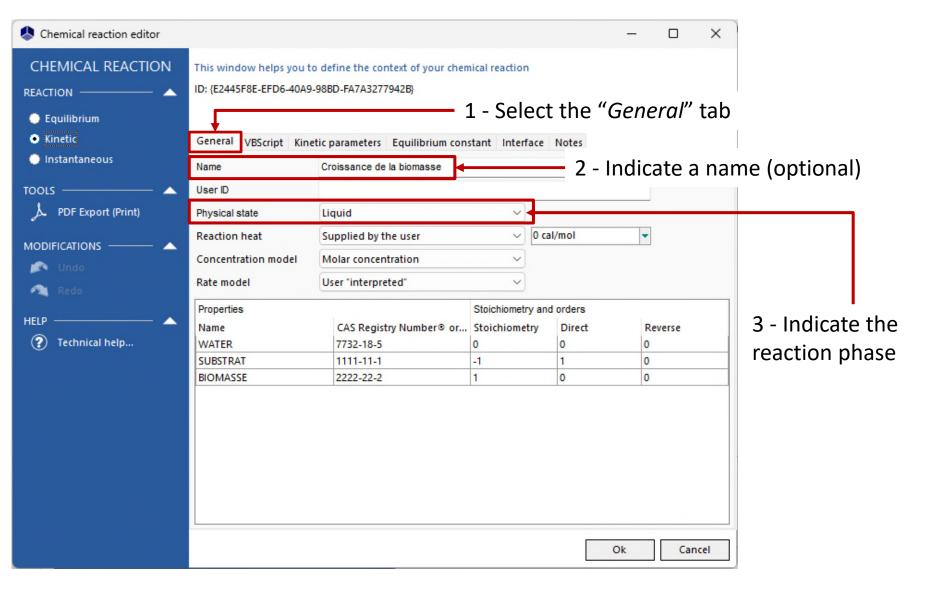
Back on the "Calculators editor" window, click on "Edit the chemical reactions of this calculator"

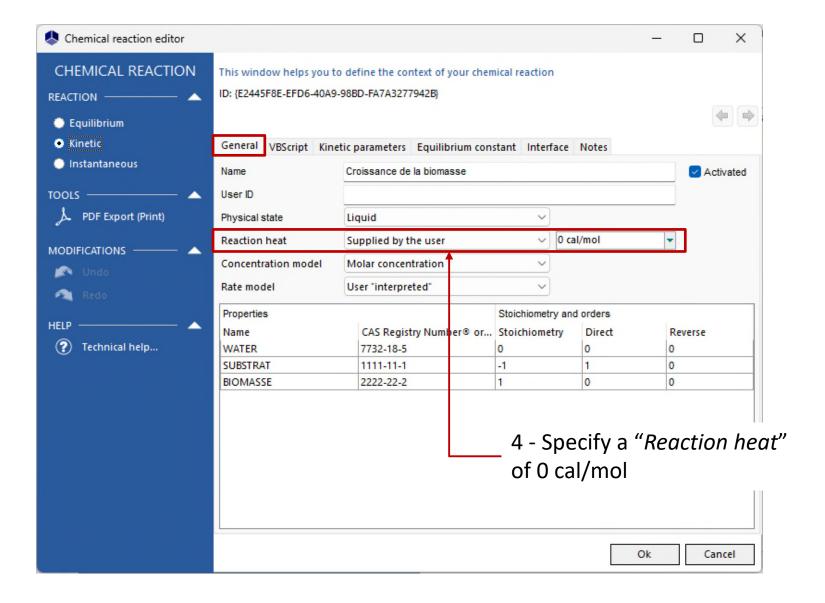


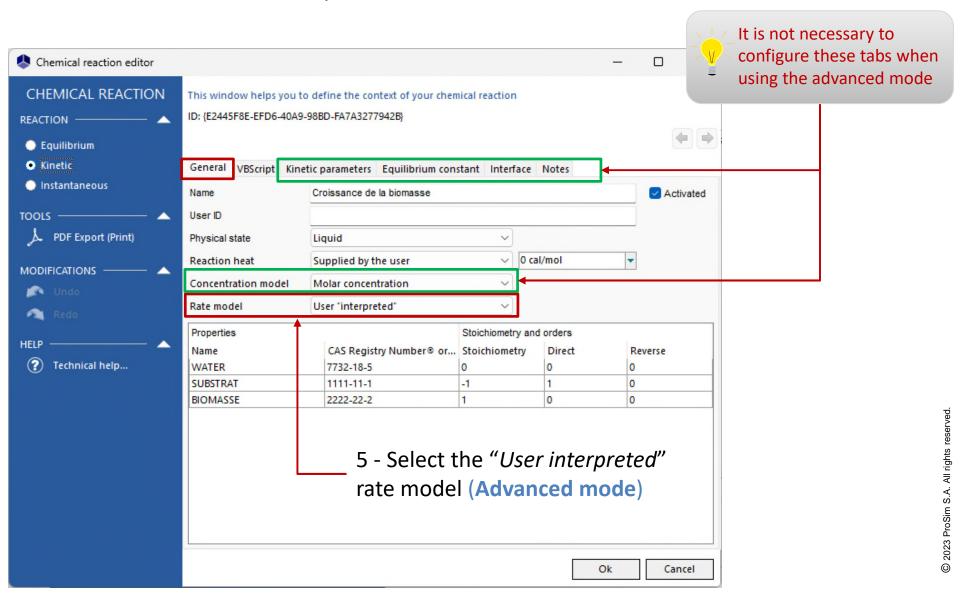
1 - Select "Add a reaction"

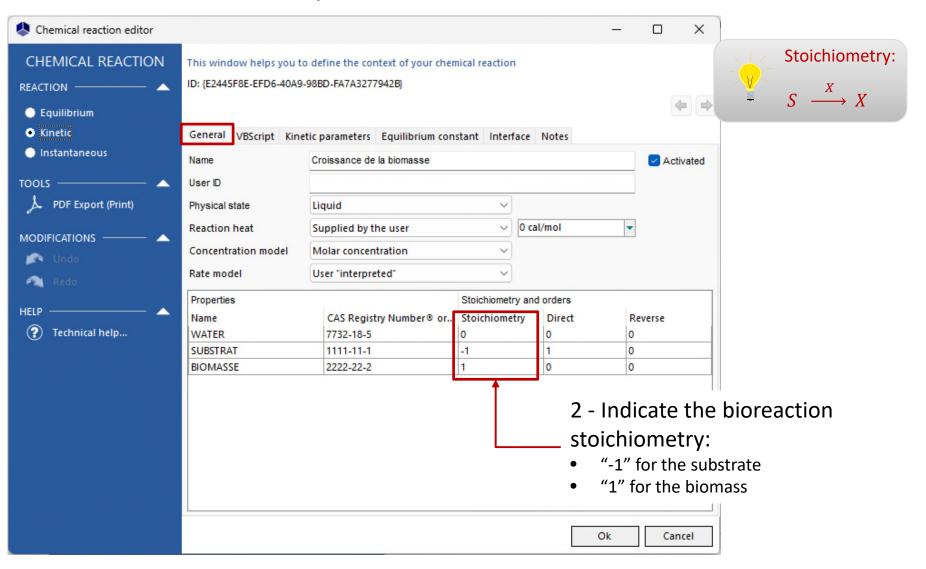


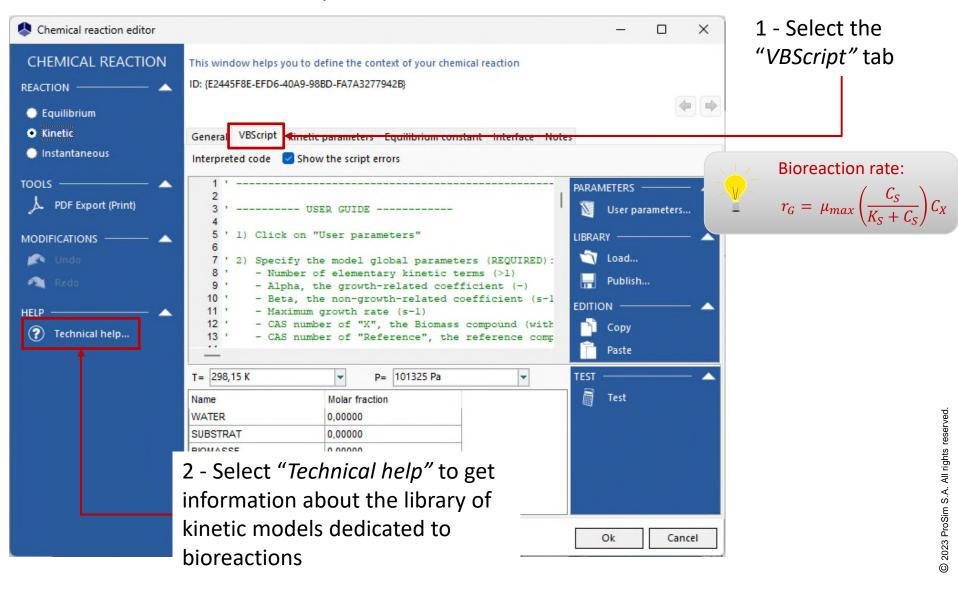
3 - Double click on the new reaction to configure it











Access the **Advanced mode** to provide a custom kinetic model for the bioreaction



It is necessary to look at the "Technical help" to get information about the suitable kinetic models for the bioreaction along with the parameters to provide

Selection of the suitable model



Bioreaction rate:

$$r_G = \mu_{max} \left(\frac{C_S}{K_S + C_S} \right) C_X$$

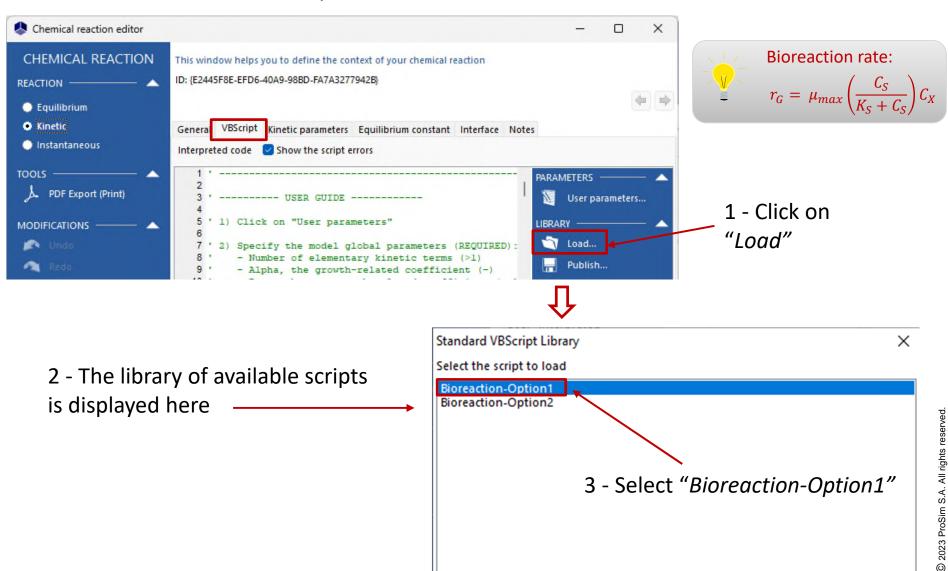
Selection of the model for the global rate of bioreaction

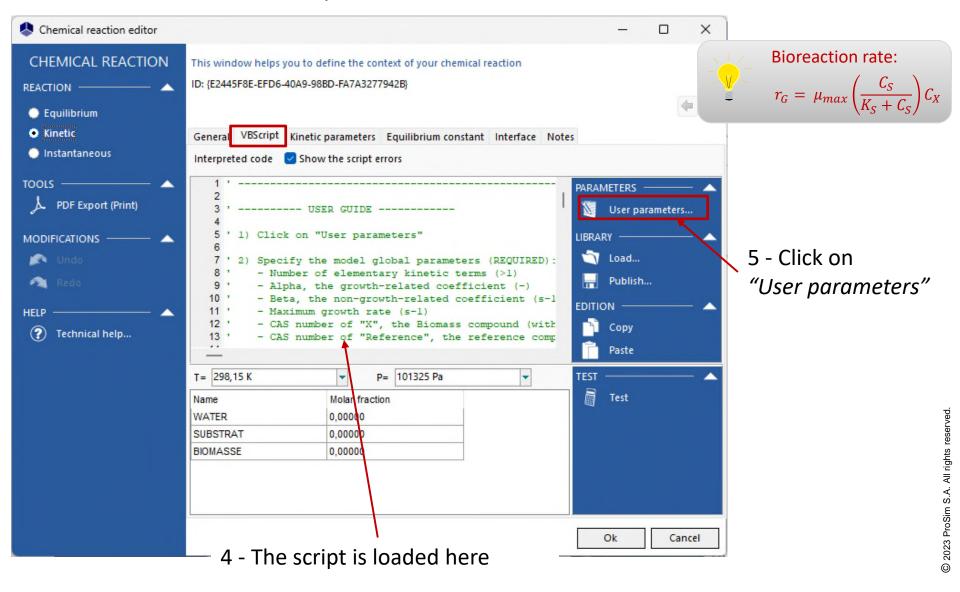
(Option 1):
$$r_G = \left(\alpha. \mu_{max} \prod_{i=1}^{NLS} r(C_{Si}) + \beta\right). C_X$$

Selection of the elementary kinetic model $r(C_{Si})$

1 elementary term

Model index	Description	Equation of the term $r(\mathcal{C}_{\mathit{Si}})$
1	Monod	$\frac{C_S}{K_S + C_S}$





Bioreaction rate:

 $r_G = \mu_{max} \left(\frac{C_S}{K_S + C_S} \right) C_X$

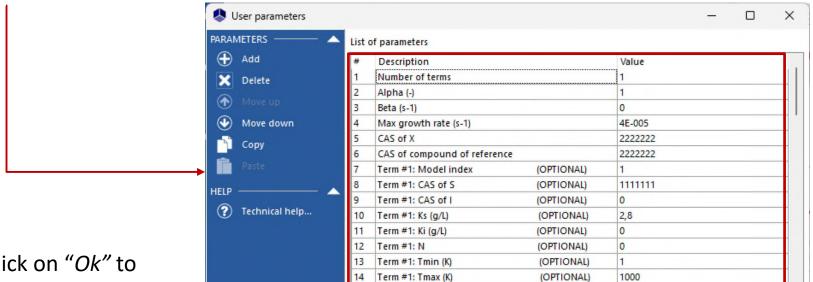
Cancel

Step 2: Configuration of the bioreaction

Access the **Advanced mode** to provide a custom kinetic model for the bioreaction

1 - Specify the model input parameters:

- 1 elementary kinetic term, corresponding to the model #1 (Monod)
- α = 1
- $\beta = 0$
- μ_{max} = 4.10⁻⁵ s⁻¹
- $K_{\rm S} = 2.8 \, {\rm g/L}$
- CAS number^(*) of the biomass (X) and the compound of reference: 2222222
- CAS number^(*) of the substrate (S): 1111111



Term #2: Model index

(OPTIONAL)

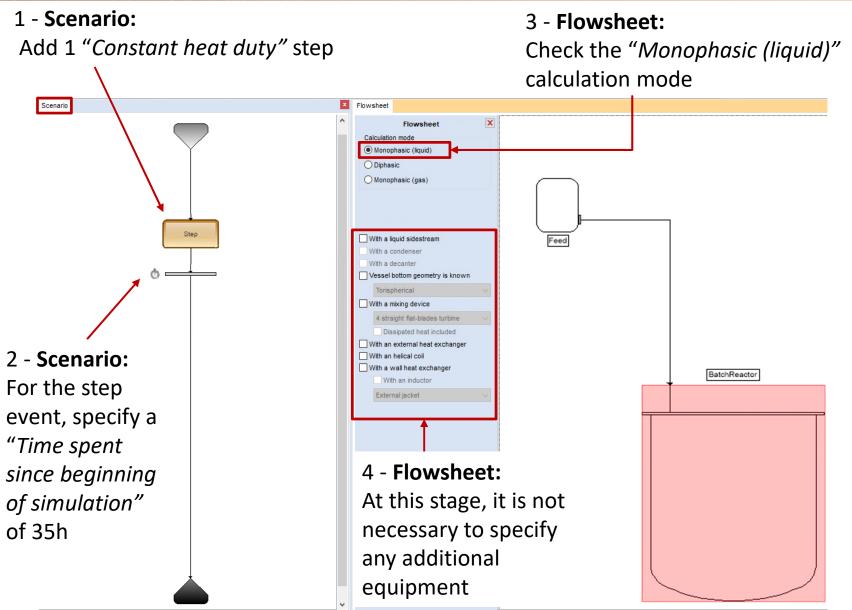
0

2 - Click on "*Ok*" to confirm and go back to the main flowsheet

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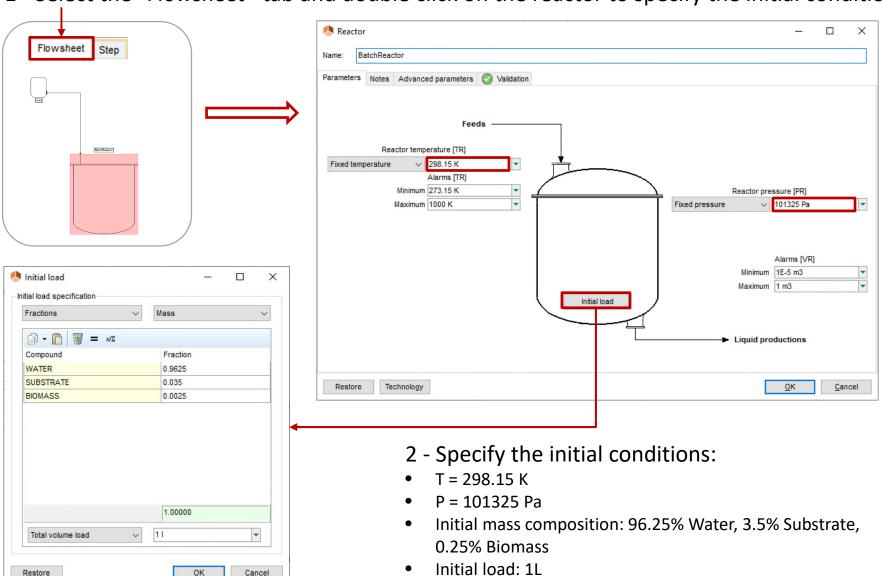
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Step 3: Specification the equipment and the operating scenario



Step 3: Specification the equipment and the operating scenario

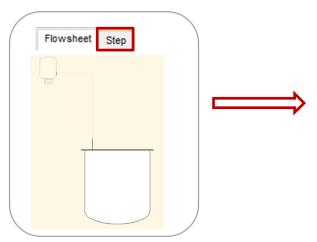
1 - Select the "Flowsheet" tab and double click on the reactor to specify the initial conditions



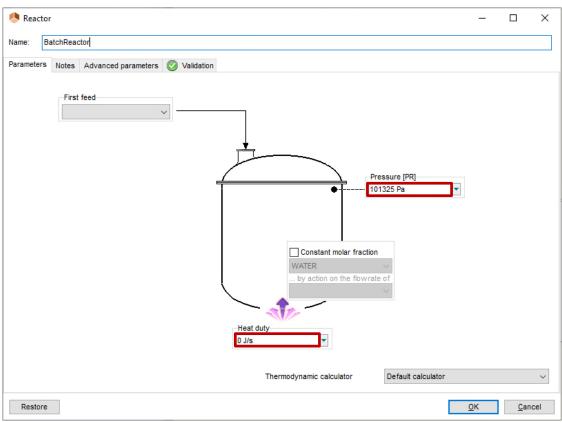
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Step 3: Specification the equipment and the operating scenario

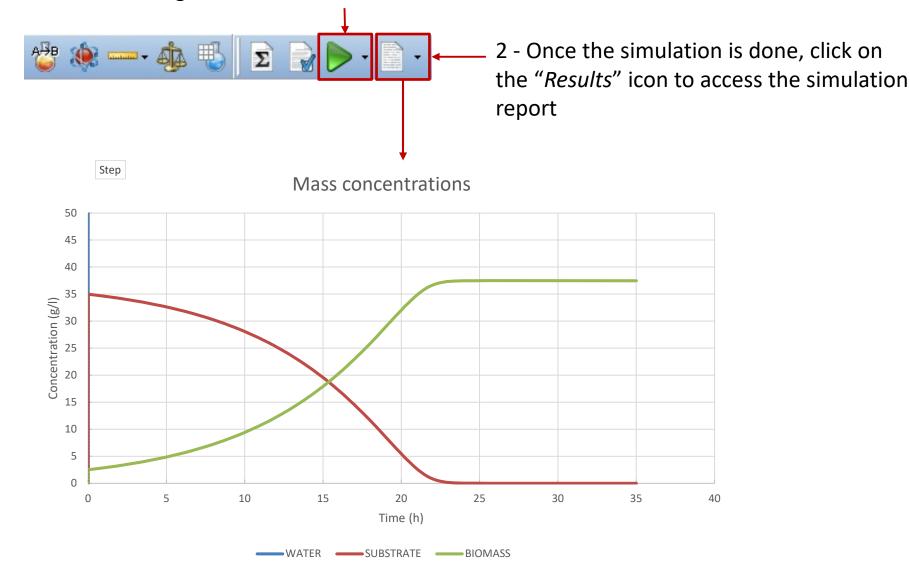
1 - Double click on the "Step" in the operating scenario or select the "Step" tab and double click on the reactor to specify the operating conditions



- 2 Specify the operating conditions:
- No feed
- Heat duty = 0 J/s
- Pressure = 101325 Pa













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